

Temperature Dependence of Three-Body Hydrophobic Interactions: Potential of Mean Force, Enthalpy, Entropy, Heat Capacity, and Nonadditivity [*J. Am. Chem. Soc.* **2005**, *127*, 303–316]. Maria Sabaye Moghaddam, Seishi Shimizu, and Hue Sun Chan*

Pages 314–315: Part of our discussion of the $T \approx 298$ K results in Table 2 needs to be corrected. This concerns four sentences: “In fact, one anti-cooperative aspect of these simulation results appears to be robust across the different studies: In all four cases, [...] this observation implies that the interaction cannot be cooperative at both the contact minimum and the desolvation barrier” (starting at the bottom of the right column on page 314 through the end of the same paragraph on page 315). These sentences should be deleted, because the observation that “the free energy difference between the two-methane contact minimum and desolvation barrier is larger than one-half of the corresponding free energy difference for three methanes” does not by itself guarantee either cooperativity or anti-cooperativity. This correction does not affect the rest of the analysis and the simulation results presented, nor does it modify the conclusions of the article.

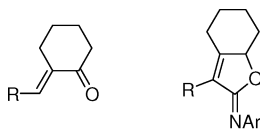
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A Catalytic [4 + 1] Cycloaddition of α, β -Unsaturated Carbonyl Compounds with Isocyanides [*J. Am. Chem. Soc.* **2005**, *127*, 761–766]. Masayuki Oshita, Kohei Yamashita, Mamoru Tobisu, and Naoto Chatani*

Page 764. Compounds **33** to **40** should be replaced with the following:



33 R = Bu^t

35 R = Pr^t

37 R = Bu

39 R = Me

34 77%

36 62%

38 53%

40 44%

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